

Gas-phase fragmentation of metallo-supramolecular aggregates

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Introduction

- Electrospray (ESI) Mass Spectrometry: an indispensable tool for the characterization of metallo-supramolecular aggregates.
- Typical signal series for the aggregates in different charge states are observed after stripping off varying number of anions.
- Stoichiometry (metal/ligand ratio) is directly determined.



- in-source fragmentation

Conclusions

- Fragmentation is strongly dominated by the charge state: lower charged complexes: ligand expulsion higher charged aggregates:
 - charge separation by cleavage into smaller aggregates, additional redox pathways with special ligands
- Enlarging the structure by elongating the ligands can change the fragmentation pattern due to a reduction of charge density.

- \succ Question:
 - Structural/topological characterization by mass spectrometry
- Analysis of fragmentation patterns in the gas phase variables: aggregate charge, size, ligand and metal type

Charge

- The charge state dominates the fragmentation
- more than the structure:
- low charge states: ligand expulsion down to salt-bridged ions high charge states: charge separation into smaller aggregates



- Unsymmetrical cleavages typically occur in such a way that high charge densities are avoided. Put the anions on the smaller fragment!
- Ligands may not be redox-innocent.
- The **topology** of the aggregate might be determined, but it is not the decisive factor for fragmentation.

Ligand Size

Smaller ligands / higher charge density change the fragmentation pattern to charge separation pathways.







Typical fragmentation pattern for dinuclear rhombs: symmetric cleavage for even number of anions. Unsymmetric cleavagage: Anions fill free metal binding sites.

Paracyclophane-based unusual ligands



Paracyclophanes: Ligand rupture into halves needs less energy than the metal-ligand bond. These ligands are **not redox-innocent**.





