

Studies of a *L*-proline catalyzed Diels-Alder reaction with inverse electron demand by ESI-MS

A. Schnell, J. A. Willms, M. Engeser

anneschnell@uni-bonn.de, marianne.engeser@uni-bonn.de Kekulé-Institute of Organic Chemistry and Biochemistry, University of Bonn

Diels-Alder reaction with inverse electron demand

In 2008, Xie et al.^[1] published a L-proline catalyzed Diels-Alder reaction with inverse electron demand between ketones and aryl-1,2,4,5-tetrazines which gives easy access to substituted functionalized pyridazines. A mechanism was postulated alongside studies of the scope of the reaction.^[1]

Conclusion

The postulated mechanism of the *L*-proline catalyzed Diels-Alder reaction with inverse electron demand was thoroughly verified. All three intermediates were detected and characterized.^[6]

Charge-tagging

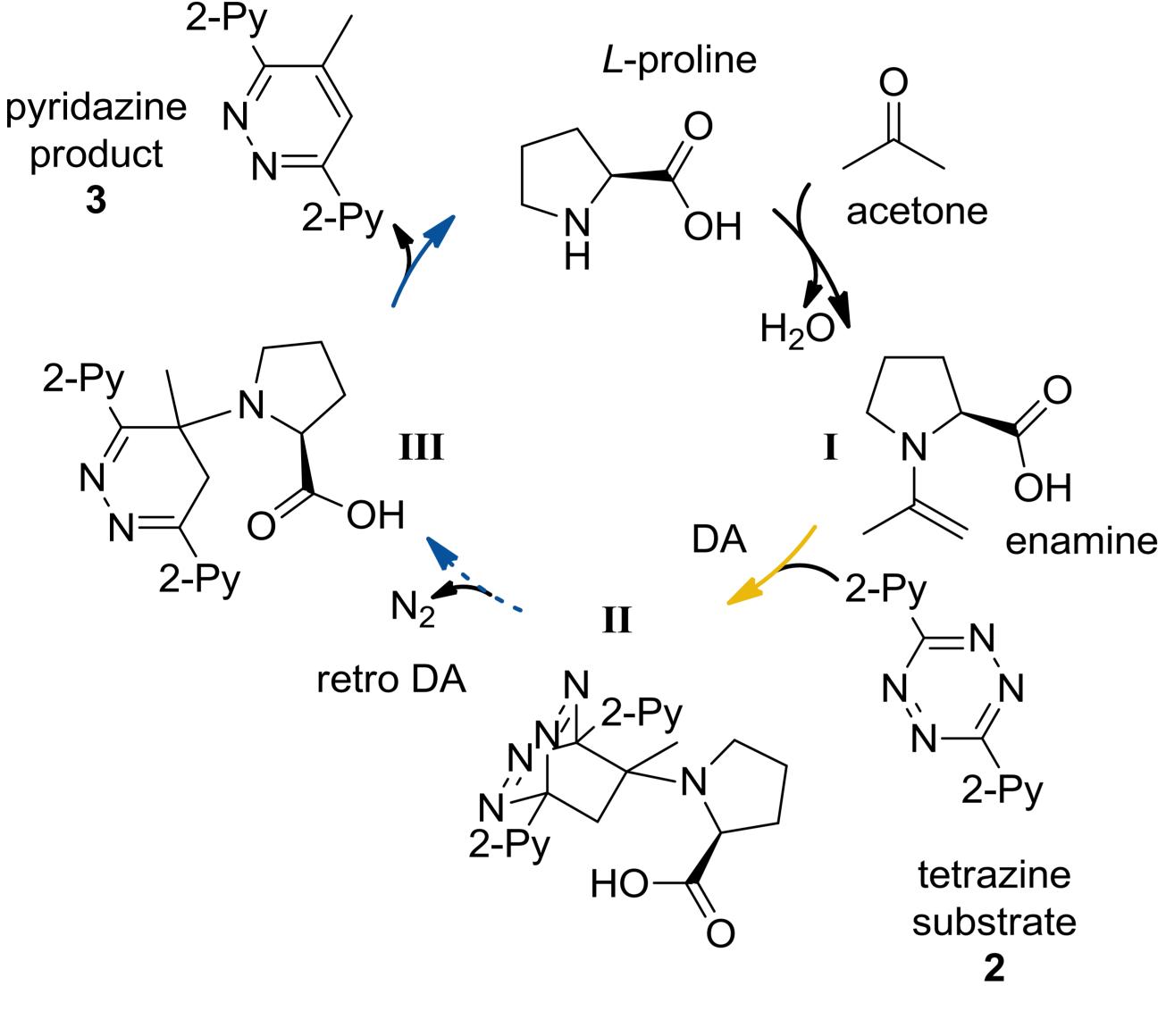
Covalently linked charge-tags can be attached to the participating

Organic

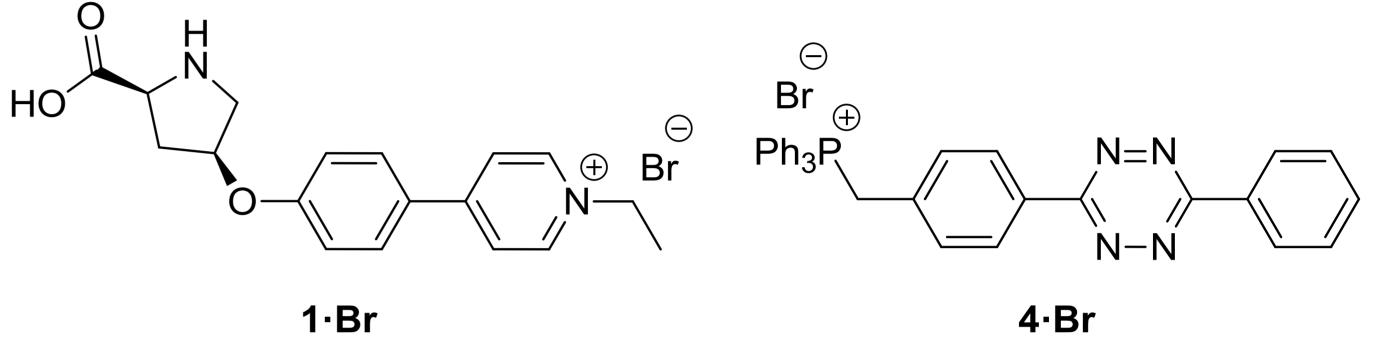
of

Institute

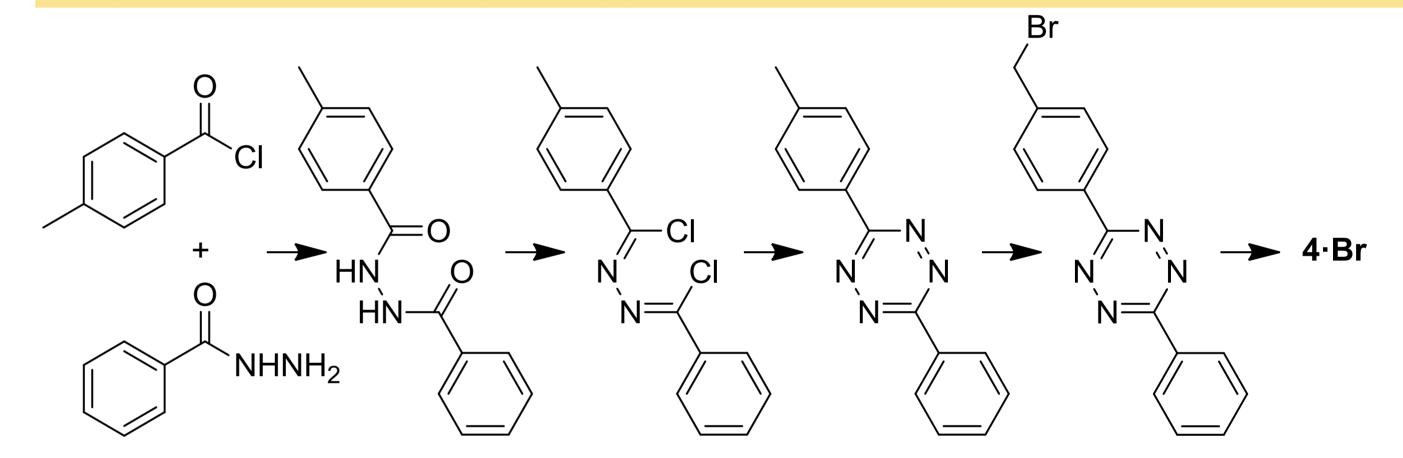
Kekulé-



molecules usually in the form of alkylated amines or phosphanes.^[2,3] As a result, all species containing the charge-tag should have similarly high ESI response factors^[2,4]. For this study the charge-tagged tetrazine **4**·**Br** was synthesized. The charge-tagged *L*-proline derivative **1**·**Br**^[5] was used as well.





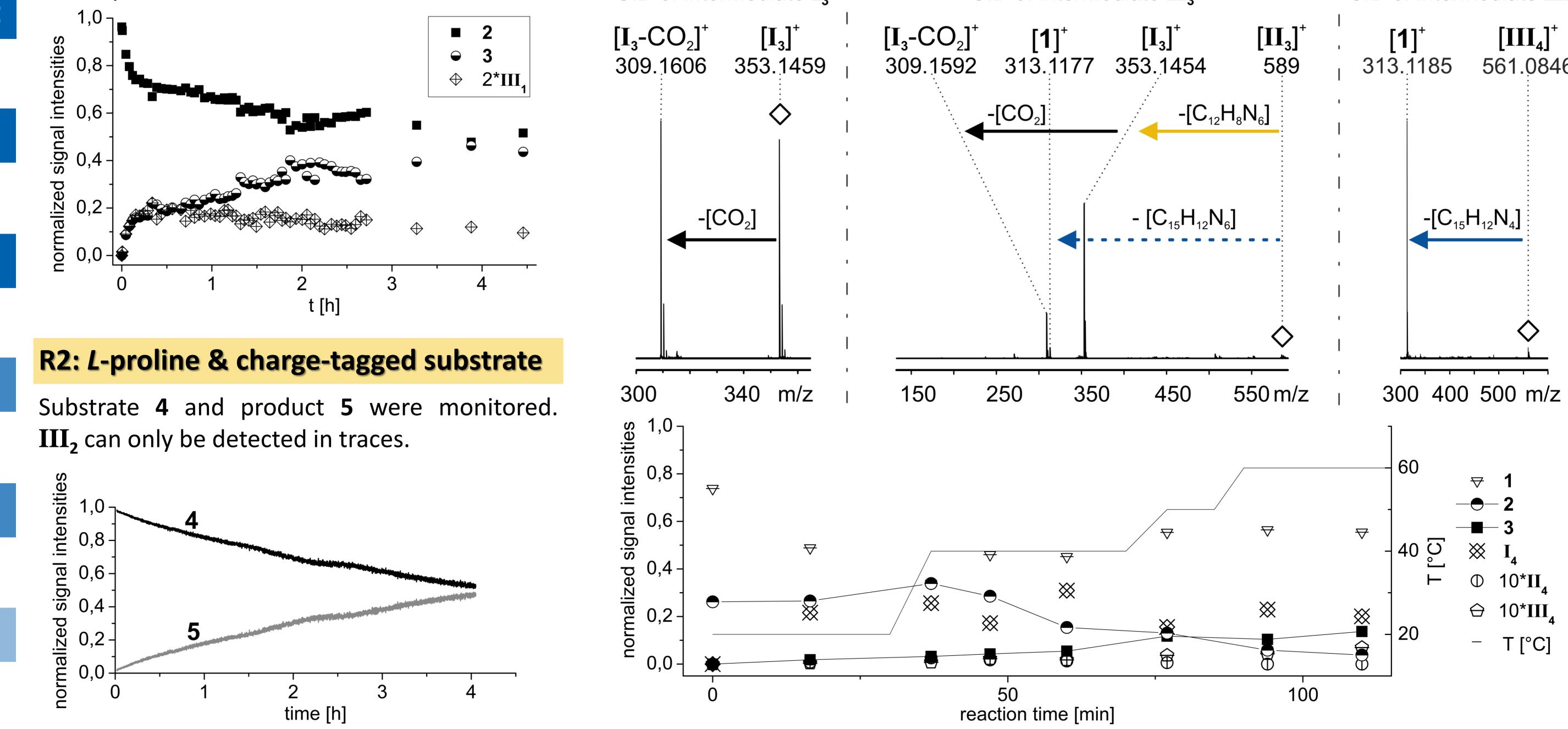


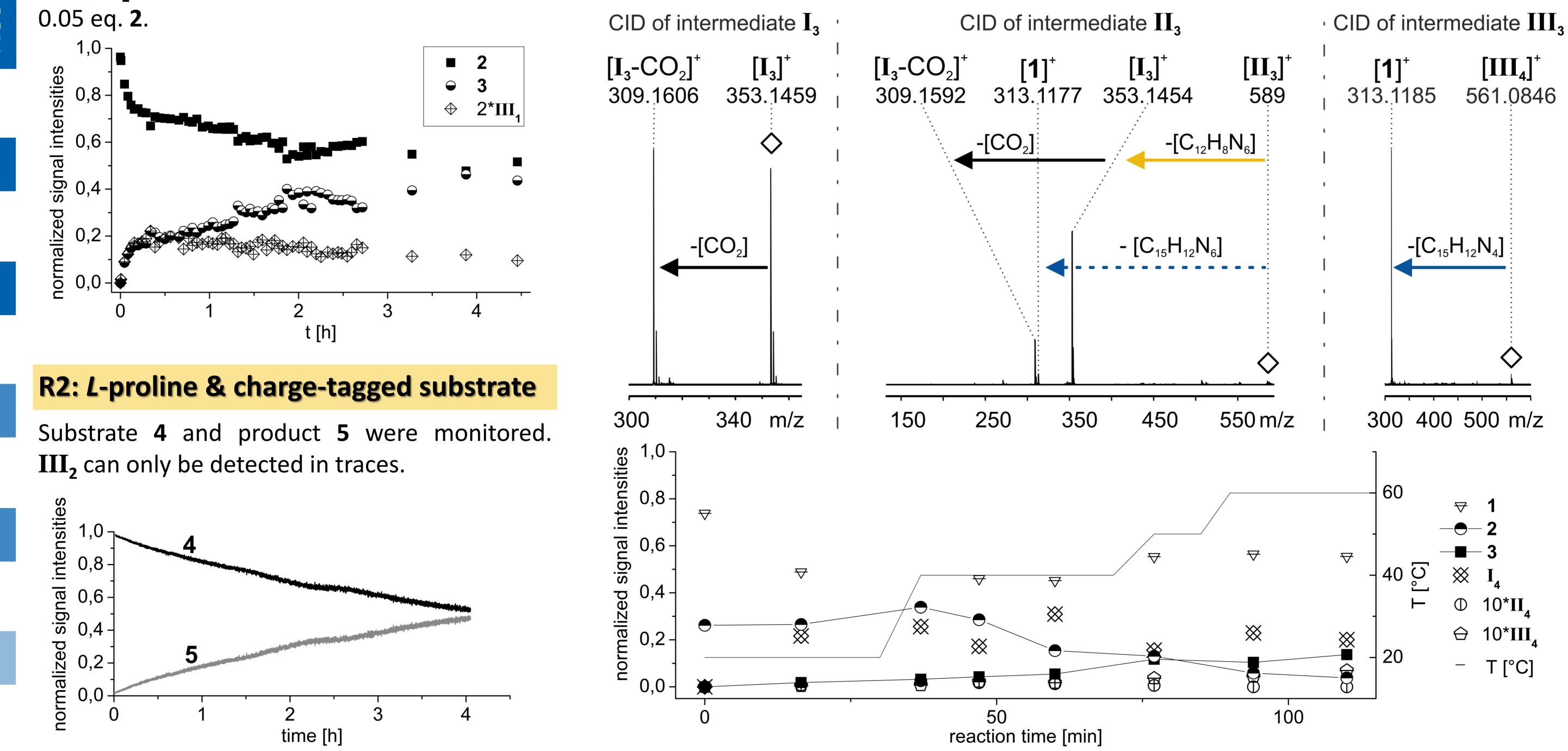
R1: *L*-proline & untagged substrate

The substrate 2, product 3, and the third intermediate **III**₁ were detected during a kinetic study. I_1 could be detected in a reaction with

R3: charge-tagged proline 1·Br & untagged substrate

Substrate 2, product 3, catalyst 1.Br, and all three intermediates were detected. CID experiments for all intermediates were conducted. Thus, single steps of the catalytic cycle could be mimicked in the gas phase.





References

- H. Xie, L. Zu, H. R. Oueis, H. Li, J. Wang, W. Wang, Org. Lett. 2008, 10, 1923–1926. [1]
- [2] L. P. E. Yunker, R. L. Stoddard, J. S. McIndoe, J. Mass Spectrom. 2014, 49, 1–8.
- [3] D. Schröder, Acc. Chem. Res. 2012, 45, 1521–1532.

- J. Luo, A. G. Oliver, J. S. McIndoe, *Dalton Trans.* **2013**, *42*, 11312–11318. [4]
- J. A. Willms, R. Beel, M. L. Schmidt, C. Mundt, M. Engeser, Beilstein J. Org. Chem. [5] **2014**, *10*, 2027–2037.
- [6] A. Schnell, J. A. Willms, M. Engeser, Beilstein J. Org. Chem. 2018, submitted.