Donor-Acceptor Complex Formation by Social Self-sorting of Polycyclic Aromatic Hydrocarbons and Perylene Bisimides

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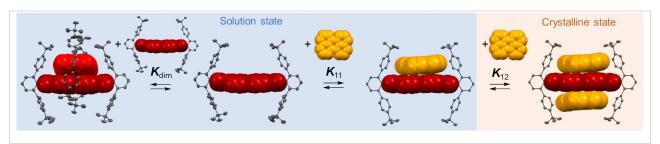


Figure 1: Perylene bisimides equipped with bulky imide substituents self-assemble into orthogonally stacked dimers or bind up to two polycyclic aromatic hydrocarbons.

Electron-donor–acceptor (EDA) complexes between electron-rich and electron-poor □-scaffolds are desired in the solid state for various applications^[1] and even more in solution for fundamental photophysical studies.^[2] However, EDA complexes between the electron-poor perylene bisimides (PBIs) and electron-rich polycyclic aromatic hydrocarbons (PAHs) could so far only be realized in solution with PBI cyclophane supramolecular hosts.^[3]

Here, self-assembly versus complexation with PAH guest molecules is studied for a series of PBIs in solution as well as in the solid state. Bulky imide substituents at the chromophore guide their self-assembly into discrete dimer aggregates with null-type exciton coupling. Host-guest titration experiments with different PAHs afford 1:1 and 1:2 complexes whose properties are studied by single crystal X-ray analysis as well as UV/Vis and fluorescence spectroscopy.^[4]

References:

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